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study about the  
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composition of

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structure,  
properties, and  
reactions of  
substances is  
called  
chemistry.

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main branch of  
science and it

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who are learning  
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Match each item  
with the correct  
statement below.

- a. calorimeter
- d. temperature
- b. calorie e.
- specific heat c.
- joule f. heat



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\_\_\_\_\_ 1. quantity  
of heat needed  
to raise the  
temperature of 1  
g of water by 1  
C \_\_\_\_\_ 2. SI  
unit of energy  
\_\_\_\_\_ 3. energy  
transferred  
between 2  
objects because  
of temperature  
difference \_\_\_\_\_  
4.

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Matter SECTION 3  
SHORT ANSWER

Answer the  
following  
questions in the  
space provided.

1. Match  
description on  
the right to the  
correct crystal  
type on the  
left. b ionic

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Crystal (a) has mobile electrons in the crystal c  
covalent  
molecular  
crystal (b) is hard, brittle, and nonconducting a  
metallic crystal  
(c) typically has the lowest melting point of the four

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(Groups) In this  
topic, Student

should be able

to : Discuss the  
variation in the

physical



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properties of  
elements  
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belonging to  
period 2 and 3  
and to describe  
and explain the  
periodicity in  
the following  
physical  
properties of  
elements.

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nuclei used to  
make an atom  
bomb: Answer:  
Uranium,

Plutonium. 34) A  
gaseous mixture  
contains oxygen  
and nitrogen in  
the ratio of 1:  
4 by weight.  
Therefore the  
ratio of their  
number of  
molecules is.

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(a) 1: 4 (b) 1:  
8 (c) 7: 32 (d)  
3: 16. Answer:  
C.

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