

Chemfax Designing A Hand Warmer

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One type of hand warmer contains a mixture of dry chemicals, including among other things powdered iron and salt. When exposed to air and moisture the iron oxidizes. The oxidation reaction is exothermic and gives off a gentle heat to warm hands.

~~AP Chemistry: Designing an Effective Hand Warmer~~

The purpose of this advanced inquiry lab is to design an effective hand warmer that is inexpensive, nontoxic and safe for the environment. The investigation begins with an introductory activity to become familiar with the principles of calorimetry and heat of solution calculations.

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Designing a Hand Warmer Lab. Purpose Statement. The purpose of this lab is to investigate the energy changes accompanying the formation of. solutions for common laboratory salts, and then apply the results to design a hand warmer that is. reliable, safe, and inexpensive. Partners. Sana Navaid. Madeline Crowley. Michelle Shumiatsky. Introduction. 1.

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Designing a Hand Warmer Purpose: To determine the best solute to use to make a safe hand

warmer. Procedure: Data Part A: (table) Part B and C: (table) Calculations: Part A: #13-

calculate calorimeter constant Part B and C: #7 Calculate the molar heat of solution q_{solution}

$= -(q_{\text{aq}} + q_{\text{cal}})$... Designing a Hand Warmer Purpose: To determine the best ...

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AP Chemistry Designing a Hand Warmer Lab Rebecca Poliner. ... Insta Heat Reusable Hand

Warmers VERSUS HotHands Air Activated Hand Warmers - Duration: 8:42. Peter von Panda

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Recommended for ...

~~AP Chemistry Designing a Hand Warmer Lab~~

Designing a Hand Warmer Lab Introduction: This lab will require you to put your chemistry skills to commercial use! From instant cold packs to flameless ration heaters and hand warmers, the energy changes accompanying physical and chemical transformations have many consumer applications. The backbone of these applications is

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To find the best solid to create an effective hand warmer a calorimeter was used. The solid with the highest endothermic reaction with 33° created, was LiCl which had a cost of \$68.30 per kilogram. Since this lab was made to discover a cheap costing solid, LiCl does not fit the requirments. LiCl is too expensive.

~~Designing a Hand Warmer by Xochitl Benitez—Prezi~~

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Siena Joy for our AP Chemistry class. Designing a Hand Warmer Lab Designing a Hand
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